

Pharmacy Math Practice Problems with Solutions

30 PTCB-style calculation problems, each with a step-by-step worked solution.

Source: ptcbquizprep.com. Solutions show every step. Last reviewed July 15, 2026.

1. A pharmacist asks you to repackage 2.5 L of an antacid suspension into unit-of-use bottles. How many milliliters are in 2.5 L?

- A. 250 mL
- B. 2,500 mL
- C. 25,000 mL
- D. 0.25 mL

Answer: B. Step 1: Recall the metric conversion $1 \text{ L} = 1,000 \text{ mL}$. Step 2: Multiply: $2.5 \text{ L} \times 1,000 \text{ mL/L} = 2,500 \text{ mL}$. Watch for decimal-shift errors: 250 mL comes from multiplying by 100 instead of 1,000, and 25,000 mL comes from multiplying by 10,000.

2. A prescription is written for 10 mL of cephalexin suspension by mouth three times daily. Using household measures, how many teaspoonfuls should the patient take per dose?

- A. 1/2 teaspoonful
- B. 1 teaspoonful
- C. 2 teaspoonfuls
- D. 3 teaspoonfuls

Answer: C. Step 1: Recall the household conversion $1 \text{ teaspoonful (tsp)} = 5 \text{ mL}$. Step 2: Divide the dose volume by 5 mL/tsp: $10 \text{ mL} / 5 \text{ mL per tsp} = 2 \text{ teaspoonfuls per dose}$. The 1/2 tsp answer comes from inverting the ratio (5/10), a common setup error.

3. A patient weighs 176 lb. What is the patient's weight in kilograms? (1 kg = 2.2 lb)

- A. 80 kg
- B. 88 kg
- C. 387.2 kg
- D. 76 kg

Answer: A. Step 1: To convert pounds to kilograms, divide by 2.2 (because $1 \text{ kg} = 2.2 \text{ lb}$). Step 2: $176 \text{ lb} / 2.2 \text{ lb per kg} = 80 \text{ kg}$. Multiplying instead of dividing gives the impossible 387.2 kg; 88 kg comes from dividing by 2 instead of 2.2.

4. An older prescription calls for aspirin gr v (5 grains). Using the exact apothecary conversion 1 gr = 64.8 mg, how many milligrams is this dose?

- A. 300 mg
- B. 64.8 mg
- C. 12.96 mg
- D. 324 mg

Answer: D. Step 1: The apothecary grain has been defined as exactly 64.8 mg since 1958. Step 2: Multiply the number of grains by 64.8 mg/gr: $5 \text{ gr} \times 64.8 \text{ mg} = 324 \text{ mg}$. Note: 300 mg results from using the rougher 60 mg/gr approximation, and 12.96 mg comes from dividing 64.8 by 5 instead of multiplying.

5. Amoxicillin oral suspension is available as 250 mg/5 mL. How many milliliters must be measured to deliver a 400 mg dose?

- A. 8 mL
- B. 5 mL
- C. 10 mL
- D. 12.5 mL

Answer: A. Step 1: Set up a proportion: $250 \text{ mg} / 5 \text{ mL} = 400 \text{ mg} / x \text{ mL}$. Step 2: Cross-multiply: $250x = 400 \times 5 = 2,000$. Step 3: Solve: $x = 2,000 / 250 = 8 \text{ mL}$. A quick check: the concentration is 50 mg/mL, and $400 / 50 = 8 \text{ mL}$.

6. An IV bag contains heparin 25,000 units in 250 mL of D5W. How many units of heparin are contained in 40 mL of this solution?

- A. 2,500 units
- B. 400 units
- C. 4,000 units
- D. 6,250 units

Answer: C. Step 1: Find the concentration: $25,000 \text{ units} / 250 \text{ mL} = 100 \text{ units/mL}$. Step 2: Multiply by the volume: $100 \text{ units/mL} \times 40 \text{ mL} = 4,000 \text{ units}$. The 6,250-unit answer comes from inverting the ratio ($25,000 / 4$), and 400 units is a decimal-shift error.

7. A benzalkonium chloride solution is labeled 1:200 (w/v). How many milligrams of benzalkonium chloride are in 15 mL of this solution?

- A. 7.5 mg
- B. 75 mg
- C. 750 mg
- D. 30 mg

Answer: B. Step 1: A 1:200 (w/v) ratio strength means 1 g of drug per 200 mL of solution. Step 2: Convert to mg: $1 \text{ g} = 1,000 \text{ mg}$, so the concentration is $1,000 \text{ mg} / 200 \text{ mL} = 5 \text{ mg/mL}$. Step 3: Multiply by the volume: $5 \text{ mg/mL} \times 15 \text{ mL} = 75 \text{ mg}$. The 7.5 mg and 750 mg answers are decimal-shift errors.

8. A stock solution contains 1.2 g of active ingredient in 480 mL. How many milliliters of this stock solution will provide a 75 mg dose?

- A. 18.75 mL
- B. 12 mL
- C. 300 mL
- D. 30 mL

Answer: D. Step 1: Convert grams to milligrams: $1.2 \text{ g} = 1,200 \text{ mg}$. Step 2: Find the concentration: $1,200 \text{ mg} / 480 \text{ mL} = 2.5 \text{ mg/mL}$. Step 3: Divide the dose by the concentration: $75 \text{ mg} / 2.5 \text{ mg per mL} = 30 \text{ mL}$. Forgetting the g-to-mg conversion or inverting the concentration produces the wrong answers.

9. How many grams of dextrose are contained in a 500 mL bag of D5W (5% dextrose in water)?

- A. 5 g
- B. 25 g
- C. 50 g
- D. 2.5 g

Answer: B. Step 1: A 5% w/v solution contains 5 g of dextrose per 100 mL. Step 2: Set up a proportion: $5 \text{ g} / 100 \text{ mL} = x \text{ g} / 500 \text{ mL}$. Step 3: Solve: $x = 5 \times (500 / 100) = 25 \text{ g}$. The 50 g answer doubles the correct value (a common slip when scaling 100 mL to 500 mL), and 5 g forgets to scale at all.

10. How many grams of sodium chloride are required to compound 1 L of normal saline, 0.9% (w/v) sodium chloride solution?

- A. 0.9 g
- B. 4.5 g
- C. 9 g
- D. 90 g

Answer: C. Step 1: 0.9% w/v means 0.9 g of NaCl per 100 mL. Step 2: Convert 1 L to 1,000 mL. Step 3: Scale up: $0.9 \text{ g} \times (1,000 / 100) = 9 \text{ g}$. Answers of 0.9 g and 90 g are decimal-shift errors; 4.5 g would be correct only for 500 mL.

11. A 45 g tube contains hydrocortisone 2.5% (w/w) cream. How many grams of hydrocortisone are in the tube?

- A. 1.125 g
- B. 11.25 g
- C. 0.1125 g
- D. 2.5 g

Answer: A. Step 1: A 2.5% w/w preparation contains 2.5 g of drug per 100 g of product. Step 2: Multiply by the total weight: $(2.5 / 100) \times 45 \text{ g} = 0.025 \times 45 = 1.125 \text{ g}$. The 11.25 g and 0.1125 g answers are decimal-shift errors, and 2.5 g mistakes the percent value itself for the drug content.

12. How many milliliters of isopropyl alcohol are needed to compound 480 mL of a 70% (v/v) isopropyl alcohol solution?

- A. 144 mL
- B. 33.6 mL
- C. 686 mL
- D. 336 mL

Answer: D. Step 1: A 70% v/v solution contains 70 mL of alcohol per 100 mL of final solution. Step 2: Multiply: $0.70 \times 480 \text{ mL} = 336 \text{ mL}$ of isopropyl alcohol. Note: 144 mL is the volume of the diluent (the other 30%), not the alcohol; 686 mL comes from dividing 480 by 0.70 instead of multiplying.

13. How many milliliters of a 10% stock solution are required to prepare 250 mL of a 2% solution?

- A. 25 mL
- B. 50 mL
- C. 125 mL
- D. 5 mL

Answer: B. Step 1: Use the dilution equation $C_1V_1 = C_2V_2$. Step 2: Plug in: $10\% \times V_1 = 2\% \times 250 \text{ mL}$. Step 3: Solve: $V_1 = (2 \times 250) / 10 = 500 / 10 = 50 \text{ mL}$ of stock, then q.s. with diluent to 250 mL. Inverting the concentrations ($10 \times 250 / 2$) would give 1,250 mL, and 25 mL comes from halving instead of solving the equation.

14. A compounding order calls for 120 g of a 5% ointment. The pharmacy stocks a 10% ointment and a 2% ointment. Using alligation, how many grams of the 10% ointment are required?

- A. 75 g
- B. 60 g
- C. 45 g
- D. 36 g

Answer: C. Step 1: Set up the alligation grid with 10% (high), 2% (low), and 5% (desired). Step 2: Parts of 10% = $5 - 2 = 3$ parts; parts of 2% = $10 - 5 = 5$ parts; total = 8 parts. Step 3: Grams of 10% ointment = $120 \text{ g} \times (3/8) = 45 \text{ g}$. Choosing 75 g mixes up the parts (that is the amount of the 2% ointment).

15. A technician dilutes 50 mL of 50% dextrose injection to a final volume of 250 mL with sterile water for injection. What is the percentage strength of the final solution?

- A. 10%
- B. 25%
- C. 5%
- D. 20%

Answer: A. Step 1: Use $C_1V_1 = C_2V_2$. Step 2: Plug in: $50\% \times 50 \text{ mL} = C_2 \times 250 \text{ mL}$. Step 3: Solve: $C_2 = 2,500 / 250 = 10\%$. A quick check: the volume increased 5-fold (50 mL to 250 mL), so the strength drops 5-fold (50% to 10%). The 25% answer comes from halving instead of using the 5-fold dilution factor.

16. How many milliliters of sterile water must be ADDED to 120 mL of a 25% solution to reduce it to a 10% solution?

- A. 300 mL
- B. 120 mL
- C. 48 mL
- D. 180 mL

Answer: D. Step 1: Find the final volume with $C_1V_1 = C_2V_2$: $25\% \times 120 \text{ mL} = 10\% \times V_2$, so $V_2 = 3,000 / 10 = 300 \text{ mL}$. Step 2: The question asks for the water ADDED, so subtract the starting volume: $300 \text{ mL} - 120 \text{ mL} = 180 \text{ mL}$. Answering 300 mL is the classic trap of reporting the final volume instead of the added diluent.

17. A prescription reads: metoprolol 50 mg, take 1 tablet by mouth twice daily, quantity dispensed #60. What is the days' supply?

- A. 15 days
- B. 30 days
- C. 60 days
- D. 45 days

Answer: B. Step 1: Calculate the daily usage: $1 \text{ tablet} \times 2 \text{ doses per day} = 2 \text{ tablets/day}$. Step 2: Divide the quantity dispensed by the daily usage: $60 \text{ tablets} / 2 \text{ tablets per day} = 30 \text{ days}$. Answering 60 days ignores the twice-daily dosing; 15 days doubles the daily use by mistake.

18. A pharmacy dispenses 150 mL of an antibiotic suspension with the directions: give 5 mL by mouth three times daily. What is the days' supply?

- A. 5 days
- B. 15 days
- C. 10 days
- D. 30 days

Answer: C. Step 1: Calculate the daily volume: $5 \text{ mL} \times 3 \text{ doses per day} = 15 \text{ mL/day}$. Step 2: Divide the total volume by the daily volume: $150 \text{ mL} / 15 \text{ mL per day} = 10 \text{ days}$. Answering 30 days uses only one 5 mL dose per day; 5 days doubles the daily volume by mistake.

19. A patient injects insulin glargine U-100, 35 units subcutaneously twice daily, using a 10 mL vial. How many full days will one vial last?

- A. 14 days
- B. 28 days
- C. 15 days
- D. 10 days

Answer: A. Step 1: U-100 insulin contains 100 units/mL, so a 10 mL vial holds $100 \times 10 = 1,000$ units. Step 2: Calculate daily use: $35 \text{ units} \times 2 = 70 \text{ units/day}$. Step 3: Divide: $1,000 / 70 = 14.28\dots$, which is 14 full days (round DOWN for days' supply; the partial 15th day cannot be completed from the vial). Answering 28 days forgets that the dose is given twice daily.

20. A child weighing 44 lb is prescribed a medication at 15 mg/kg/day divided into two equal doses. How many milligrams are in each dose?

- A. 300 mg
- B. 330 mg
- C. 75 mg
- D. 150 mg

Answer: D. Step 1: Convert weight to kg: $44 \text{ lb} / 2.2 = 20 \text{ kg}$. Step 2: Calculate the total daily dose: $15 \text{ mg/kg} \times 20 \text{ kg} = 300 \text{ mg/day}$. Step 3: Divide into two doses: $300 / 2 = 150 \text{ mg per dose}$. Answering 300 mg gives the whole day's dose, and 330 mg comes from skipping the lb-to-kg conversion ($44 \times 15 / 2$).

21. Cefdinir is prescribed at 7 mg/kg per dose every 12 hours. What is the amount of each dose for a child weighing 66 lb?

- A. 105 mg
- B. 210 mg
- C. 462 mg
- D. 420 mg

Answer: B. Step 1: Convert weight to kg: $66 \text{ lb} / 2.2 = 30 \text{ kg}$. Step 2: Multiply by the per-dose rate: $7 \text{ mg/kg} \times 30 \text{ kg} = 210 \text{ mg per dose}$. Answering 462 mg comes from using pounds instead of kilograms (66×7), and 420 mg is the total daily amount (both doses), not a single dose.

22. An order calls for a single vancomycin dose of 20 mg/kg IV for a patient weighing 187 lb. The reconstituted vancomycin solution contains 50 mg/mL. How many milliliters of the solution are needed for this dose?

- A. 17 mL
- B. 3.4 mL
- C. 34 mL
- D. 37.4 mL

Answer: C. Step 1: Convert weight to kg: $187 \text{ lb} / 2.2 = 85 \text{ kg}$. Step 2: Calculate the dose: $20 \text{ mg/kg} \times 85 \text{ kg} = 1,700 \text{ mg}$. Step 3: Divide by the concentration: $1,700 \text{ mg} / 50 \text{ mg per mL} = 34 \text{ mL}$. Answering 37.4 mL comes from using pounds instead of kilograms; 3.4 mL and 17 mL are decimal and halving errors.

23. An IV infusion of 1,000 mL of normal saline is ordered to run over 8 hours. What is the flow rate in mL/hr?

- A. 125 mL/hr
- B. 100 mL/hr
- C. 133 mL/hr
- D. 80 mL/hr

Answer: A. Step 1: Flow rate (mL/hr) = total volume / total time in hours. Step 2: $1,000 \text{ mL} / 8 \text{ hr} = 125 \text{ mL/hr}$. Answering 80 mL/hr comes from inverting the setup (8 into 1,000 misread as 1,000 into 8 with a decimal shift), and 100 mL/hr comes from dividing by 10 instead of 8.

24. One liter of D5W is to infuse over 12 hours using an IV administration set calibrated at 15 gtt/mL. What is the flow rate, rounded to the nearest whole drop per minute?

- A. 1,250 gtt/min
- B. 6 gtt/min
- C. 42 gtt/min
- D. 21 gtt/min

Answer: D. Step 1: Use the formula $\text{gtt/min} = (\text{volume in mL} \times \text{drop factor}) / \text{time in MINUTES}$. Step 2: Convert: 1 L = 1,000 mL and $12 \text{ hr} \times 60 = 720 \text{ minutes}$. Step 3: Calculate: $(1,000 \times 15) / 720 = 15,000 / 720 = 20.83$, which rounds to 21 gtt/min. Answering 1,250 gtt/min comes from dividing by 12 hours instead of 720 minutes; 42 gtt/min comes from using 360 minutes (6 hours) by mistake.

25. An IV is infusing at 30 gtt/min through an administration set calibrated at 10 gtt/mL. What is the infusion rate in mL/hr?

- A. 18 mL/hr
- B. 180 mL/hr
- C. 300 mL/hr
- D. 20 mL/hr

Answer: B. Step 1: Convert drops per minute to mL per minute by dividing by the drop factor: $30 \text{ gtt/min} / 10 \text{ gtt/mL} = 3 \text{ mL/min}$. Step 2: Convert to hours: $3 \text{ mL/min} \times 60 \text{ min/hr} = 180 \text{ mL/hr}$. Answering 300 mL/hr comes from multiplying by the drop factor instead of dividing; 18 mL/hr is a decimal-shift error; 20 mL/hr comes from inverting the drop-factor division.

26. Potassium chloride oral solution contains 20 mEq per 15 mL. How many milliliters are needed to provide a 40 mEq dose?

- A. 15 mL
- B. 7.5 mL
- C. 30 mL
- D. 60 mL

Answer: C. Step 1: Set up a proportion: $20 \text{ mEq} / 15 \text{ mL} = 40 \text{ mEq} / x \text{ mL}$. Step 2: Cross-multiply: $20x = 40 \times 15 = 600$. Step 3: Solve: $x = 600 / 20 = 30 \text{ mL}$. A quick check: the dose is exactly double the labeled 20 mEq, so double the volume ($15 \text{ mL} \times 2 = 30 \text{ mL}$). Answering 7.5 mL comes from inverting the proportion.

27. Approximately how many milliequivalents of sodium chloride (molecular weight = 58.5; valence = 1) are contained in 1 L of 0.9% sodium chloride injection?

- A. 154 mEq
- B. 15.4 mEq
- C. 9 mEq
- D. 585 mEq

Answer: A. Step 1: Find the mass of NaCl: $0.9\% \text{ w/v} = 0.9 \text{ g/100 mL}$, so 1 L (1,000 mL) contains $9 \text{ g} = 9,000 \text{ mg}$. Step 2: Use $\text{mEq} = (\text{mg} \times \text{valence}) / \text{molecular weight} = (9,000 \times 1) / 58.5$. Step 3: Calculate: $9,000 / 58.5 = 153.8$, or approximately 154 mEq. Answering 15.4 mEq is a decimal-shift error; 9 mEq confuses grams with milliequivalents.

28. A pharmacy purchases an OTC blood glucose meter for \$48.00 and applies a 25% markup on cost. What is the selling price?

- A. \$52.00
- B. \$64.00
- C. \$12.00
- D. \$60.00

Answer: D. Step 1: Calculate the markup amount: $\$48.00 \times 0.25 = \12.00 . Step 2: Add the markup to the cost: $\$48.00 + \$12.00 = \$60.00$. Answering \$12.00 gives only the markup, not the selling price; \$52.00 comes from misreading 25% as a flat \$4 or using a decimal-shift markup.

29. A wholesaler invoice totals \$250.00 and carries a 2% prompt-payment discount if paid within 15 days. How much does the pharmacy pay if it settles the invoice within the discount window?

- A. \$240.00
- B. \$245.00
- C. \$247.50
- D. \$255.00

Answer: B. Step 1: Calculate the discount: $\$250.00 \times 0.02 = \5.00 . Step 2: Subtract the discount from the invoice total: $\$250.00 - \$5.00 = \$245.00$. Answering \$247.50 uses a 1% discount, \$240.00 uses 4%, and \$255.00 adds the discount instead of subtracting it.

30. A prescription retails for \$65.00 and the drug's acquisition cost to the pharmacy is \$39.00. What is the gross profit margin expressed as a percentage of the selling price?

- A. 26%
- B. 66.7%
- C. 40%
- D. 60%

Answer: C. Step 1: Calculate the gross profit: $\text{selling price} - \text{cost} = \$65.00 - \$39.00 = \26.00 . Step 2: Divide the gross profit by the SELLING PRICE: $\$26.00 / \$65.00 = 0.40$. Step 3: Convert to a percentage: $0.40 \times 100 = 40\%$. Answering 66.7% computes markup on cost ($\$26 / \39) instead of margin on selling price; 60% is the cost-to-price ratio; 26% mistakes the dollar profit for a percentage.

